FERTILIZERS

Nitrogen up to leaves and shoots Phosphorus down into the roots Potassium for health all around

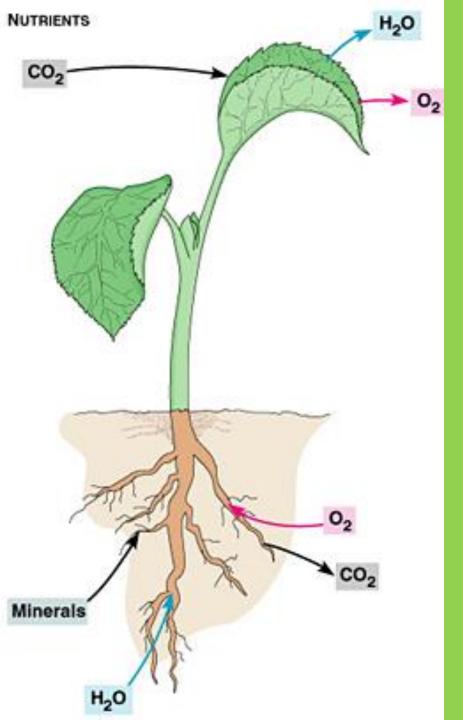
> 2019 Master Gardener Training Class Emily Gianfortoni, Hanover Master Gardener

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Life Depends on Plants

- Plants unique among almost all other organisms on earth
- Can manufacture or synthesize own food (sugars and carbohydrates) from carbon dioxide (CO2) and water (H2O) in the presence of light (sun).
- Simple sugars composed of carbon, hydrogen and oxygen



How do plants get the elements they need to photosynthesize and grow?

- Three essential elements: C, H, O
- Carbon dioxide through stomates in leaves
- Water and oxygen through roots

To grow plant needs other elements

- To produce proteins plants need nitrogen (N), sulfur (S) and other elements
- To manufacture chlorophyll need N, magnesium (Mg), and iron (Fe)
- Other elements needed for enzymes, plant hormones, and other functions

Essential Elements for Plant Growth

- Major Elements (macronutrients)
 - -N Nitrogen
 - -P Phosphorus
 - -K Potassium
 - -Ca Calcium
 - -Mg Magnesium
 - -S Sulfur



Essential Elements for Plant Growth

- Minor Elements
 - Fe Iron
 - -Cu Copper
 - -Zn Zinc
 - Mn Manganese
- * not listed in all sources

- (Micronutrients)
 - -MB -
 - Molybdenum
 - -Ni Nickel*
 - -Cl Chlorine
 - -Co-Cobalt*
 - -B Boron

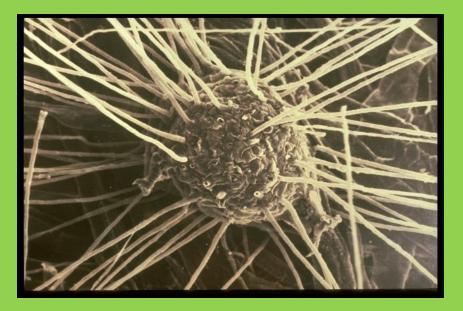
How Plants Take Up Elements From Soil

- 90% of plant
 species associated
 with root fungi
- Mycorrhizal = mutual relationship between roots and fungi



- Mycorrhizae
 penetrate root
 hairs
- Make water and nutrients more available to plant
- Receive sugars & carbohydrates from plant



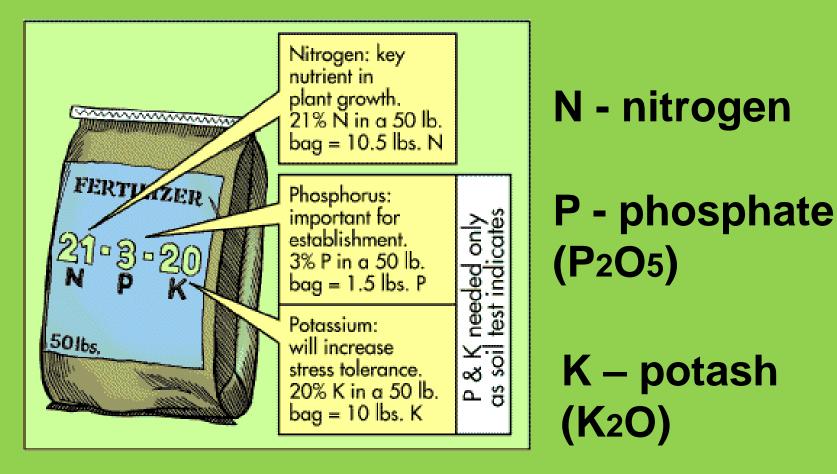


Fertilizers

Some reasons to use:

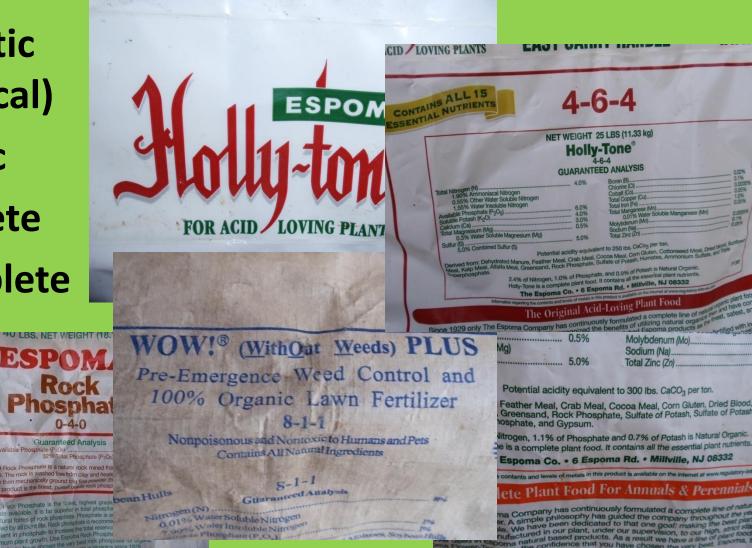
- Increase productivity of plants such as vegetables
- Improve health of plant
- Supplement elements that are low (after soil testing)

Required: prominent display of percentage by weight of 3 major elements on packaging



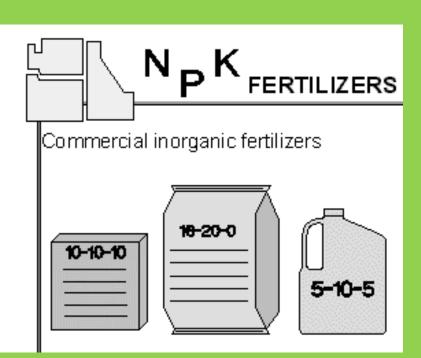
Types of Fertilizers

- Synthetic (chemical)
- Organic
- Complete
- Incomplete



Synthetic (chemical or inorganic)





Synthetic Fertilizers

- Most are in the form of salts
- Act more quickly making elements readily available
- Because are salts, can "burn" plants
- Can be detrimental to soil microorganisms
- Urea is a synthetic organic fertilizer manufactured from inorganic materials

Organic Fertilizers

Garden Safe All Purpose Natural Organic Pla	ant Food
GUARANTEED ANALYSIS	5-3-3
Total Nitrogen (N)	
1.0% Water Soluble Nitrogen	
4.0% Water Insoluble Nitrogen*	
Available Phosphate (PoOs)	
Soluble Potash (KoO)	J.U /0
Calcium (Ca)	9.0%
Derived from: Poultry manure	
"4.0% slowly available Nitrogen from poultry manure	F644



Organic Fertilizers

- Elements are from once-living organisms
- Depend on soil microorganisms to break them down and reduce to elements
- Therefore are mostly water insoluble (WIN) and break down slowly
- Have lower quantity of necessary elements than synthetic fertilizers
- Improve soil quality and increase bacteria and fungi, especially the mycorrhiza
- Micorrhizal fungi make elements more available to plants

Earthworm Castings 1-0-0

- 0.1% N (water soluble)
- 0.9% N (water insoluble)
- 0.2% Fe (iron)





Complete and Incomplete

- Complete when contains N P and K
- Incomplete contains one or two of N P or K
- Incomplete useful when soil tests show one or two of the elements to be very high and other(s) low
- Note: complete doesn't mean all major and minor elements included

Complete Synthetic: Colorburst Flowering Plant Food 15-30-15

- 11.7% ammoniacal N
- 3.3% urea (polymer coated)
- 30% available phosphate
- 15% soluble potash





Complete Organic

Neptune's Harvest

- 2-4-1 fish fertilizer
- Allowed for organic production
- Made from N.
 Atlantic fish and seaweed

Incomplete Organic Organic Seaweed Concentrate 0-4-4

- 4% phosphoric acid
- 4% soluble potash
- A soluble seaweed extract to be used as foliar spray or soil drench



Incomplete Synthetic Rot Stop: Calcium Chloride



 Sold as a treatment for blossom end rot on tomatoes

 Foliar spray to correct calcium deficiency

Minerals

- Inorganic but naturally occurring
- Examples:
 - -Greensand (source of potassium)
 - –Rock phosphate (source of phosphorus)
 - -Dolomitic limestone (calcium and magnesium)

Green sand 0-0-3 (iron-potassium silicate



- Mined from sedimentary rock deposits
- 3% potash from ironpotassium silicate
- Also includes iron, magnesium, silica and other trace minerals

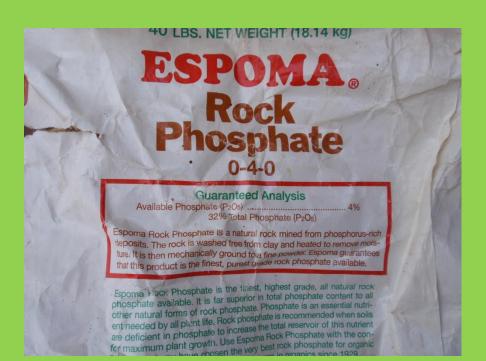
Muriate of potash 0-0-60

- 60% soluble potash
- Is the chemical compound potassium chloride
- Extracted from mineral sylvite or manufactured



Rock Phosphate 0-4-0

- Natural rock mined from phosphorus rich deposits, ground, washed to remove clay, and heated to remove moisture)
- Problem: not a renewable resource



Fertilizer Formulations

Granulated or dry

- Most common type
- Formulations sold for different types of plants, (mostly marketing); read label



Espoma Plant Tone 5-3-3

- Granular
- Derived from: hydrolyzed feather meal, pasteurized poultry manure, cocoa meal, bone meal, alfalfa meal, greensand, humates, sulfate of potash and sulfide of potash magnesium





Granular Lawn Fertilizers



<u>Liquid or water</u> <u>soluble</u>

- Best for annuals, vegetables, container plants
- Gives plant a quick boost, but not good for soil microorganisms





Miracid Soil Acidifier Plant Food 30-10-10

- Water soluble
- Also contains less than 1% of
 - B (boron)
 - Cu (chelated copper)
 - Fe (chelated iron)
 - Mn (chelated manganese)
 - Mb (molybdenum)
 - Zn (chelated zinc)



Polymer coated



-Slow release because coated with a polymer that responds to soil temperatures & water -Most slowly release elements over 3 or 4 months -Have been formulated for different types of plants (e.g. vegetables, acid loving)

Osmocote Slow Release 9-6-6

For azaleas and rhododendrons with soil acidifiers

- Ammoniacal N coated
- Nitrate N
- Phosphate coated
- Soluble potash coated
- Sulfur



Iron water soluble – coated

<u>Foliar</u>

- Soluble in water; sprayed on plant
- Elements quickly absorbed by plant
- But only to be used as supplement



Fertilizer Take Away Points If You Use Chemical Fertilizers



- Test soil before applying fertilizer
- Excess P and N in runoff pollute streams and Chesapeake Bay
- Apply correct ratio of N-P-K according to soil test
- WIN (water insoluble) and slow release best for shrubs, lawns, perennials
- Water soluble fertilizers can be used for annuals, some vegetables, containers
- Many potting soils contain slow release polymer coated fertilizer

Alternative to Synthetic Fertilizers

- Use organic WIN fertilizers
- Let the microorganisms do the work
- Read Teaming with Microbes



A Few Take Away Points



- Use compost and organic materials to improve soil and build population of bacteria and fungi
- Use synthetic fertilizers in recommended amounts only after soil testing (as salts can kill microorganisms)

- If you kill the soil bacteria and fungi, you must continue to fertilize with synthetic fertilizers
- To restore soil bacteria and fungi use compost, composted mulches and compost tea
- Humates (acids remaining at the end of composting process) can improve soil quality (example Rich Earth brand)

Before Applying Fertilizer or Amendments: Don't Guess, Soil Test



Why test your soil?

- To determine fertilizer
 needs so that correct
 type and amount can be
 applied
- To measure pH (acidity)
 of soil so that it can be
 corrected for type of
 plant to be grown

Plants that don't need annual fertilizing

- Trees: receive nutrients from fallen leaves, mulch & native soil
- Shrubs: if growth slow, top dress with compost or use balanced fertilizer at appropriate time
- Herbs: too much fertilizer or organic matter may lower content of essential oils

Plants that benefit from fertilizer

- Annuals & bedding plants: polymer coated or other slow release fertilizers, especially if dead plant material is removed in fall
- Vegetables: balanced fertilizer should be incorporated into top 6 in. of soil before planting
- House plants: use dilute liquid fertilizer only during spring & summer months

Questions?

